

Methane from Biofuels

NEIWPC Webinars

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Methane Hazards

- ❑ Methane is a flammable gas
 - Lower explosive limit (LEL) = 5%
 - Upper explosive limit (UEL) = 15%
 - Mixtures containing less than 21% oxygen may become flammable when mixed with air

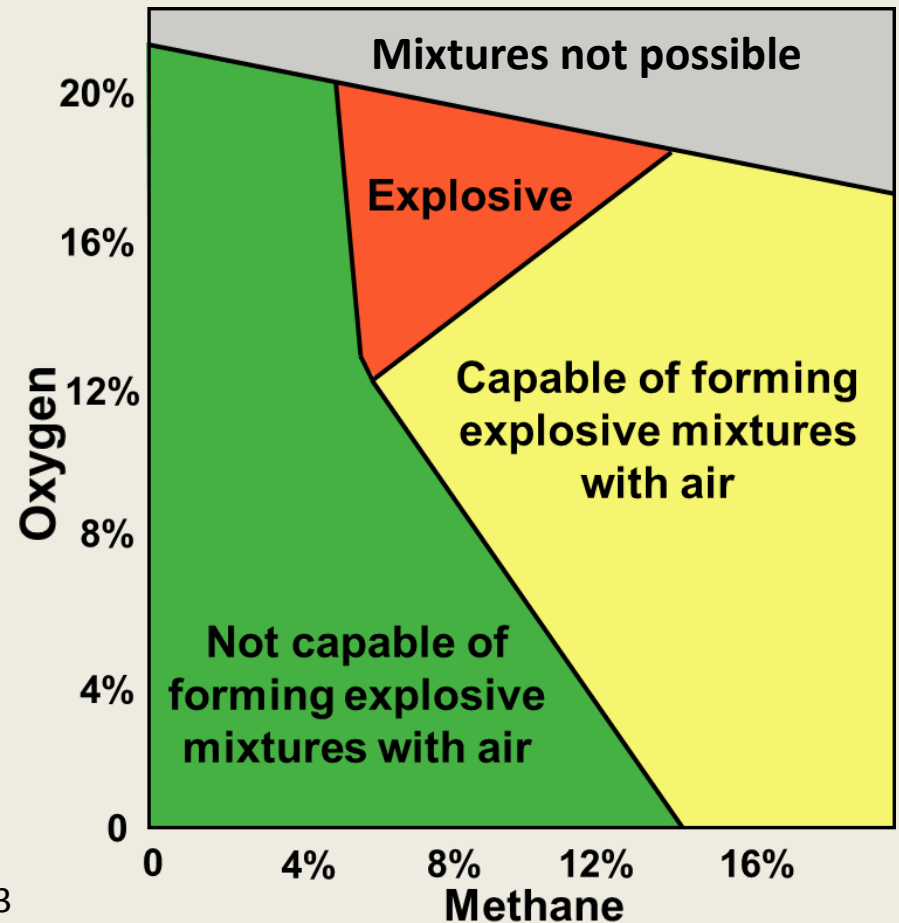


Figure source:
30 CFR 57.22003



Concerns with Biofuels

- ❑ Very high methane generation rates observed at ethanol and biodiesel release sites
 - May create immediate hazards (**methane has no odor!**)
- ❑ Different SCM than a petroleum release
 - Different analytical parameters, meters, investigation techniques
 - Longer duration monitoring
 - Additional VI monitoring may be required
- ❑ High ethanol concentrations (6% - 10%) may initially inhibit biological degradation
 - Time delay between release and methane generation possible



Findings from Case Studies

- ❑ Methane appearance may be delayed, suggesting extended monitoring if receptors are present
 - Low or no initial methane may not mean no future risk
 - May appear without detection of source biofuel
- ❑ Have seen >12% methane in a well casing headspace with aqueous-phase methane as low as 7,200 µg/L
 - Sub-saturation levels of dissolved methane can create risks
- ❑ Methane flux rates exceeding those of a Municipal Solid Waste Landfill have been observed



Findings from Case Studies

- ❑ Methane levels as high as 22% v/v in a closed surface chamber have been observed
- ❑ Biogenic gas production can lead to:
 - Stripping of petroleum VOCs from groundwater and advection of petroleum vapor by methane and other biogenic gases
 - Methane exerts a large oxygen demand which can allow petroleum vapors and methane to migrate further
- ❑ Methane in soil gas maybe the risk driver for high % biofuel blend releases



Ebullition



Surface and Flux Measurements



Flux Ring

Surface sampler
(5 gal bucket)



Monitoring Well Headspace Measurements

Landfill gas meter
(GEM)

Tubing

Stopper (toy ball)



Monitoring well headspace concentrations at a B100 Site

Well	CH ₄ %	CO ₂ %	O ₂ %	PID ppm
MW-1	0	2.9	14.6	2.0
MW-2	67.0	33.0	0	3.2
MW-3	55.3	27.4	0	4.7
MW-4	65.7	34.3	0.1	7.8
MW-5	0	3.5	9.7	1.2
MW-6	0	7.6	2.1	6.8
MW-7	38.4	25.0	0	1.6

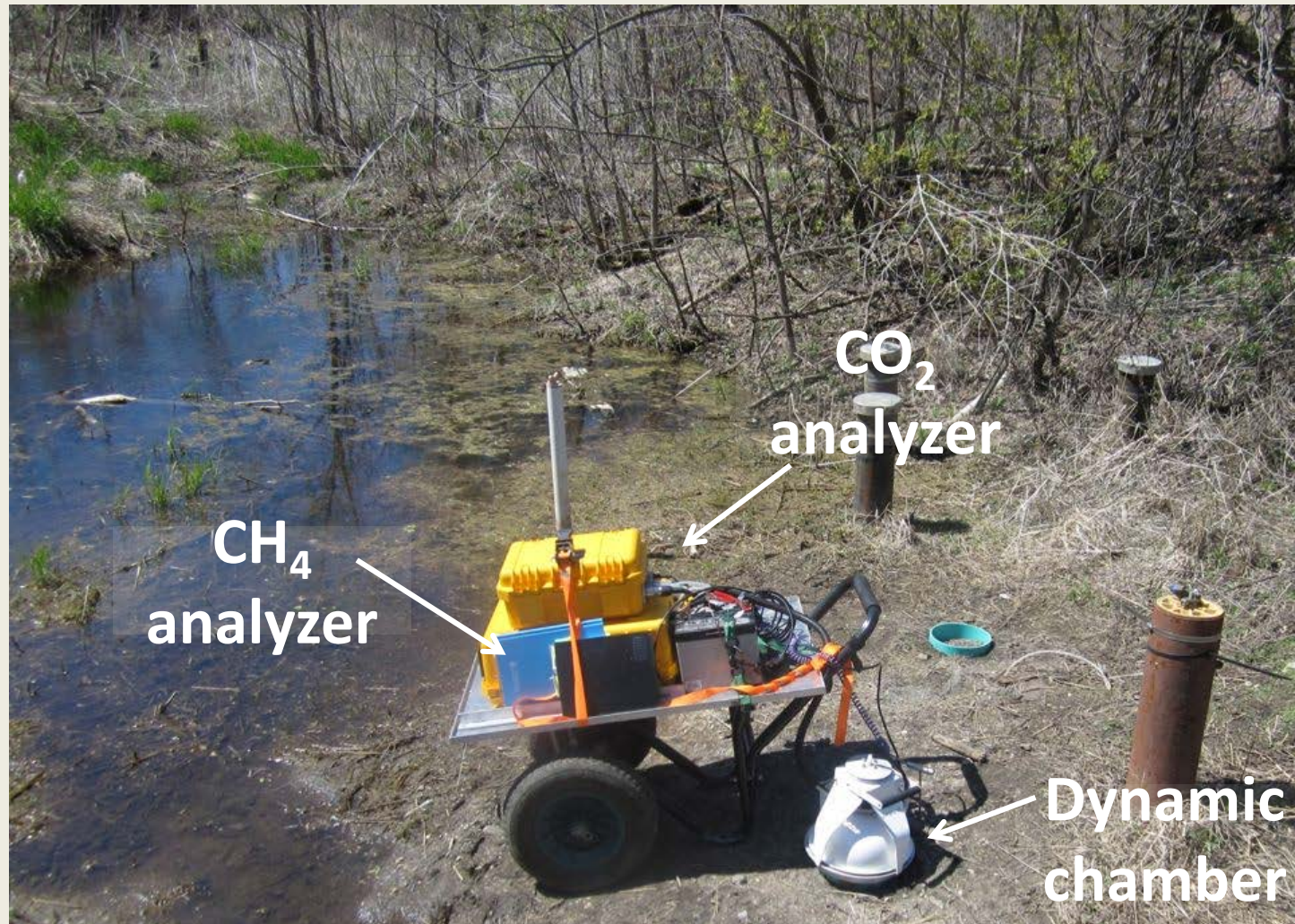


November 2006 - Cambria, MN

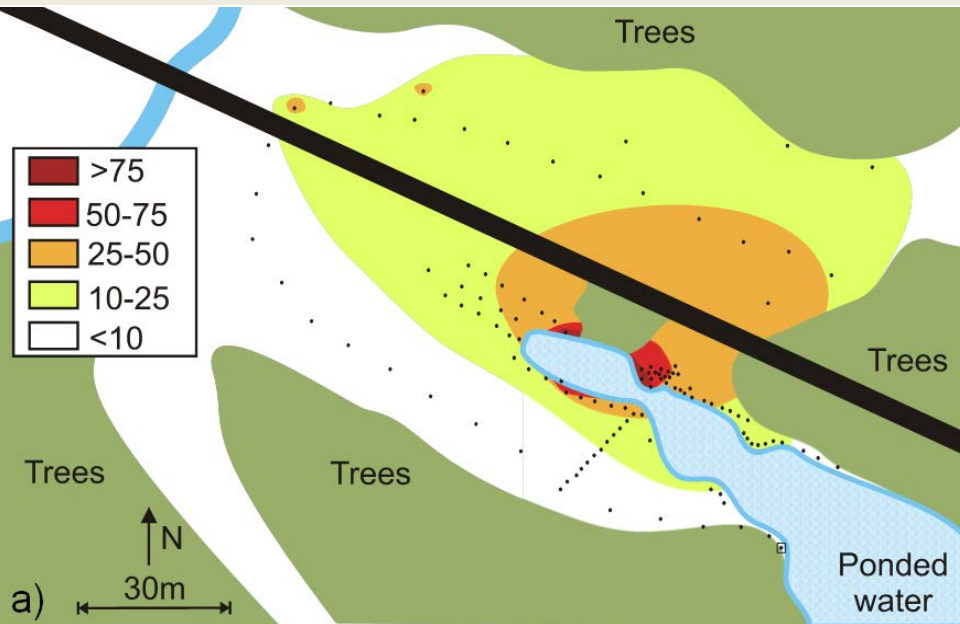
25,000 gals DFE (E95)



Real-time CO_2 and CH_4 Surficial Flux Measurements



Real Time Gas Flux Results at Cambria



a) CO₂ effluxes

b) CH₄ effluxes

[$\mu\text{mol m}^{-2} \text{s}^{-1}$]

Locally high CO₂ and CH₄ fluxes

Related to existing EtOH in capillary fringe

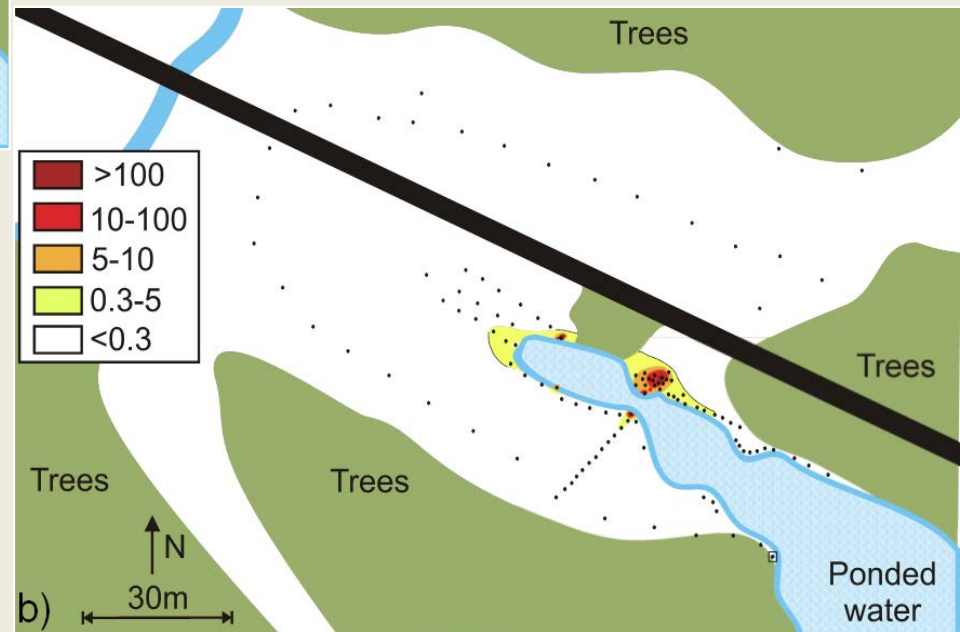


Table 2Real time CO₂ and CH₄ effluxes measured at Balaton and Cambria sites in 2012.

		Balaton	Cambria
CO ₂ efflux [$\mu\text{mol m}^{-2} \text{s}^{-1}$]	Minimum	0.6	0.5
	Maximum	178.4	174.7
CH ₄ Efflux [$\mu\text{mol m}^{-2} \text{s}^{-1}$]	Minimum	ND	ND
	Maximum	9.1	392.9
Approximate background CO ₂ efflux ^a (\pm SE, n) [$\mu\text{mol m}^{-2} \text{s}^{-1}$]		5 (1, 2)	6 (0.8, 2)
Average CO ₂ efflux attributable of DFE degradation in the source zone (\pm SE, n) [$\mu\text{mol m}^{-2} \text{s}^{-1}$]		21 (4, 55)	26 (2, 128)
Average CH ₄ efflux attributable of DFE degradation in the source zone (\pm SE, n) [$\mu\text{mol m}^{-2} \text{s}^{-1}$]		1.4 (0.5, 19)	24 (9, 46)
Total carbon flux from DFE degradation in the source zone [$\mu\text{mol m}^{-2} \text{s}^{-1}$]	Average (\pm SE, n)	22 (4, 55)	32 (5, 128)
	Minimum	5	6
	Maximum	174	500
Source Zone CH ₄ :CO ₂ ratio ^b	Average (\pm SE, n)	0.04 (0.01, 19)	0.5 (0.1, 46)
	Minimum	0.003	0.005
	Maximum	0.12	3.4 ^c
Depth-integrated DFE degradation rate in the source zone [mgEtOH m ⁻² s ⁻¹]	Average (\pm SE, n)	0.3 (0.06, 55)	0.5 (0.07, 128)
	Minimum	0.07	0.1
	Maximum	3	8

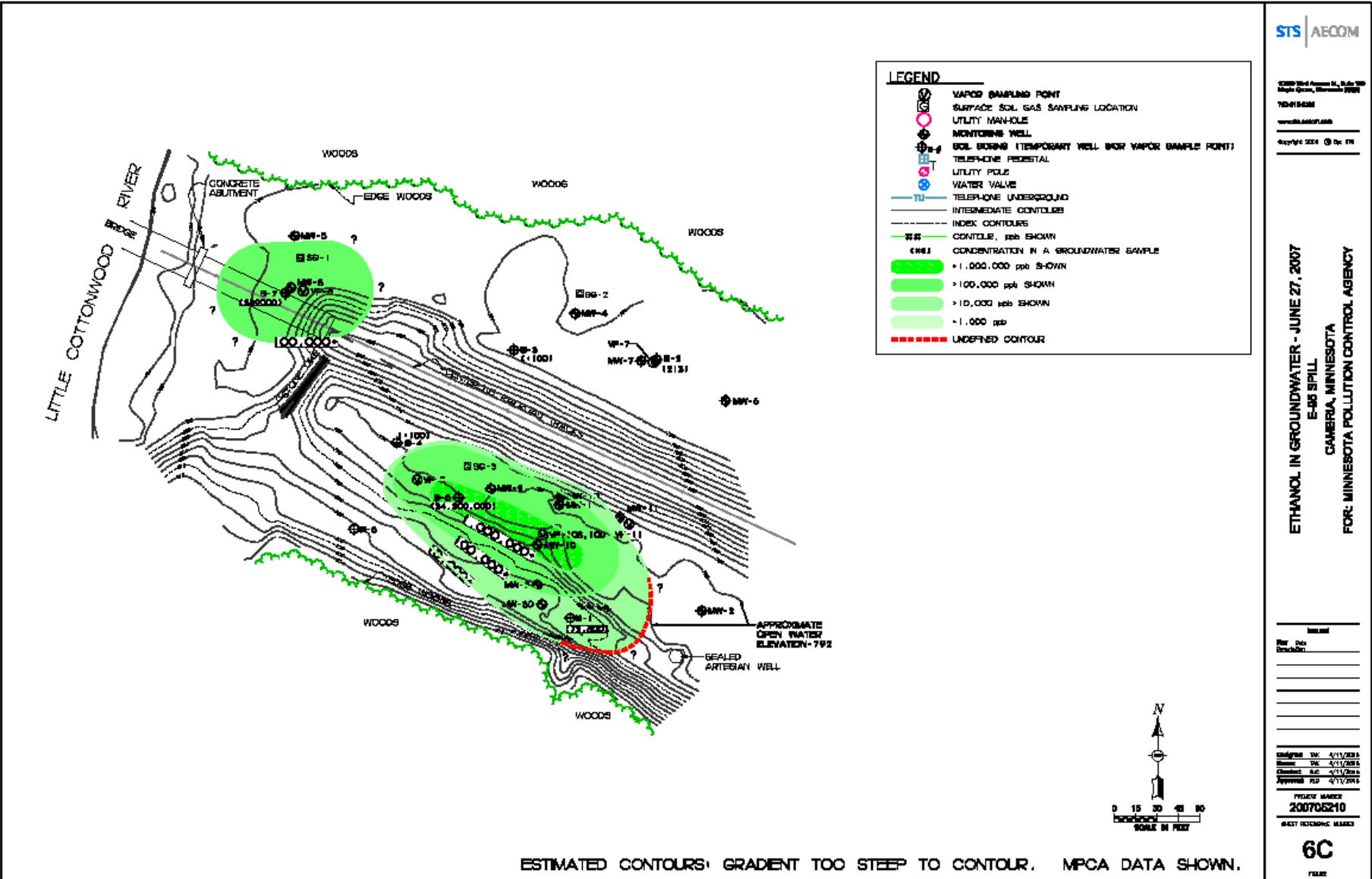
^a Within the area defined as the source zone and CH₄ fluxes were detected.^b Assumed (geometric mean of replicate measurements). ND: non-detect. BG: background CO₂ efflux correction used. SE = standard error, n = number of measurements.^c Based on an ebullition flux.

- Published active MSW landfill CH₄ flux rates: **37-94** $\mu\text{mol m}^{-2} \text{s}^{-1}$
- Cambria source zone: ave **24**, max **393** $\mu\text{mol m}^{-2} \text{s}^{-1}$
- **24** $\mu\text{mol m}^{-2} \text{s}^{-1}$ = **50 L of methane gas per m² per day!**



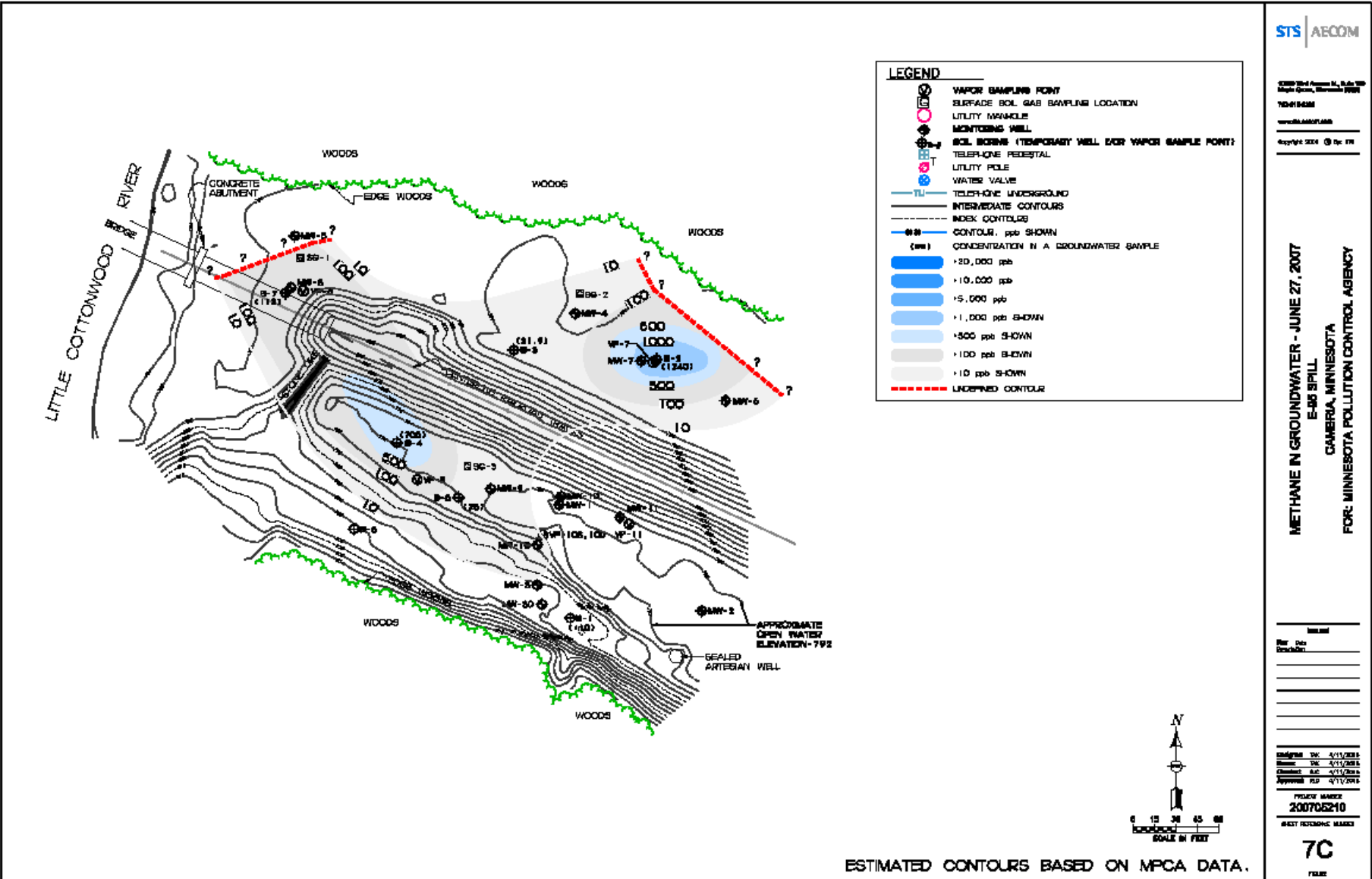
Cambria: Ethanol in Ground Water 6/07

X:\PROJECTS\200705210 BASE DRAWINGS\200705210-CAMB-ETH-METH-05.dwg, FIG. 8C, 9/24/2008 2:07:16 PM, \$TIS_PLOT\$TAMP, \$TIS_PLOT\$TAMP



Cambria: Methane in Ground Water 6/07

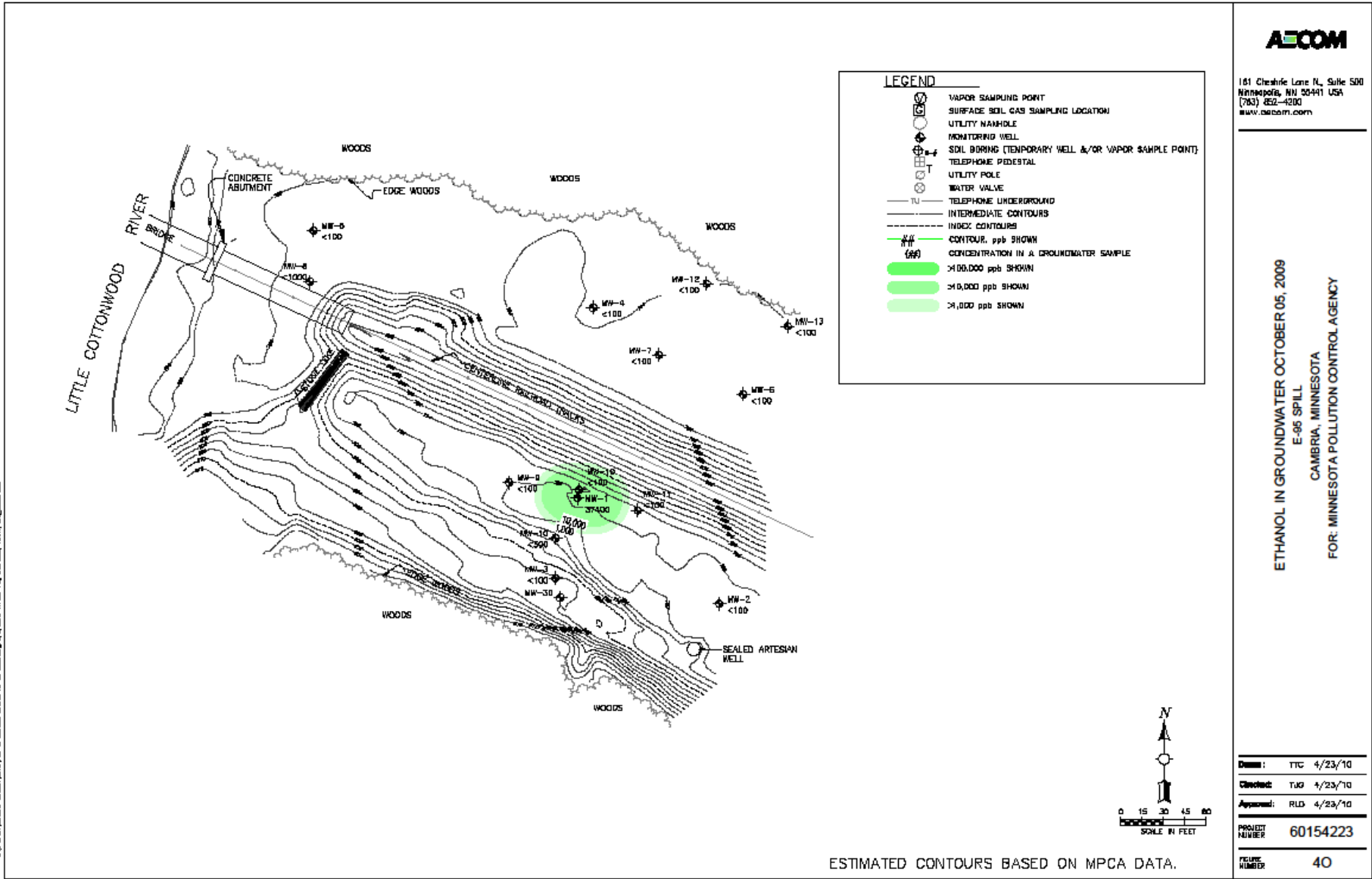
X:\PROJECTS\20060407\200705210 BASE DRAWINGS\200705210-CAMB-ETH-METH-05.dwg, FIG. 7C, 9/24/2008 4:43:04 PM, \$T\$.PLOT\$TAMP, \$T\$.PLOT\$TAMP



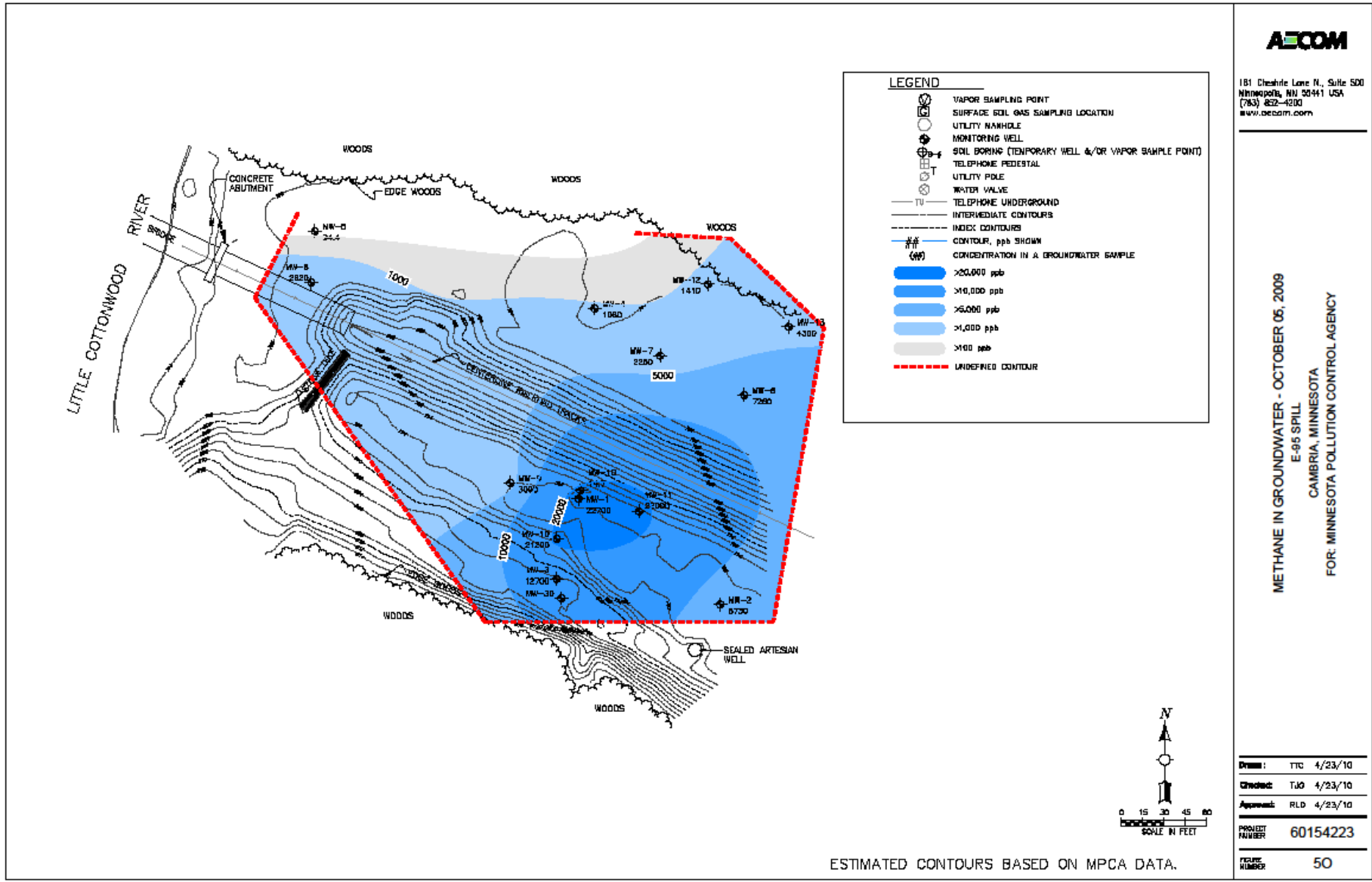
K:\PROJECTS\20090407\200705210 BASE DRAWING\3S\200705210-CAMIBETH-METH-05.dwg, FIG. 7D, 9/24/2008 1:43:40 PM, \$T8_PLOT8TAMP, \$T8_PLOT8TAMP



Cambria: Ethanol in Ground Water 10/09



Cambria: Methane in Ground Water 10/09

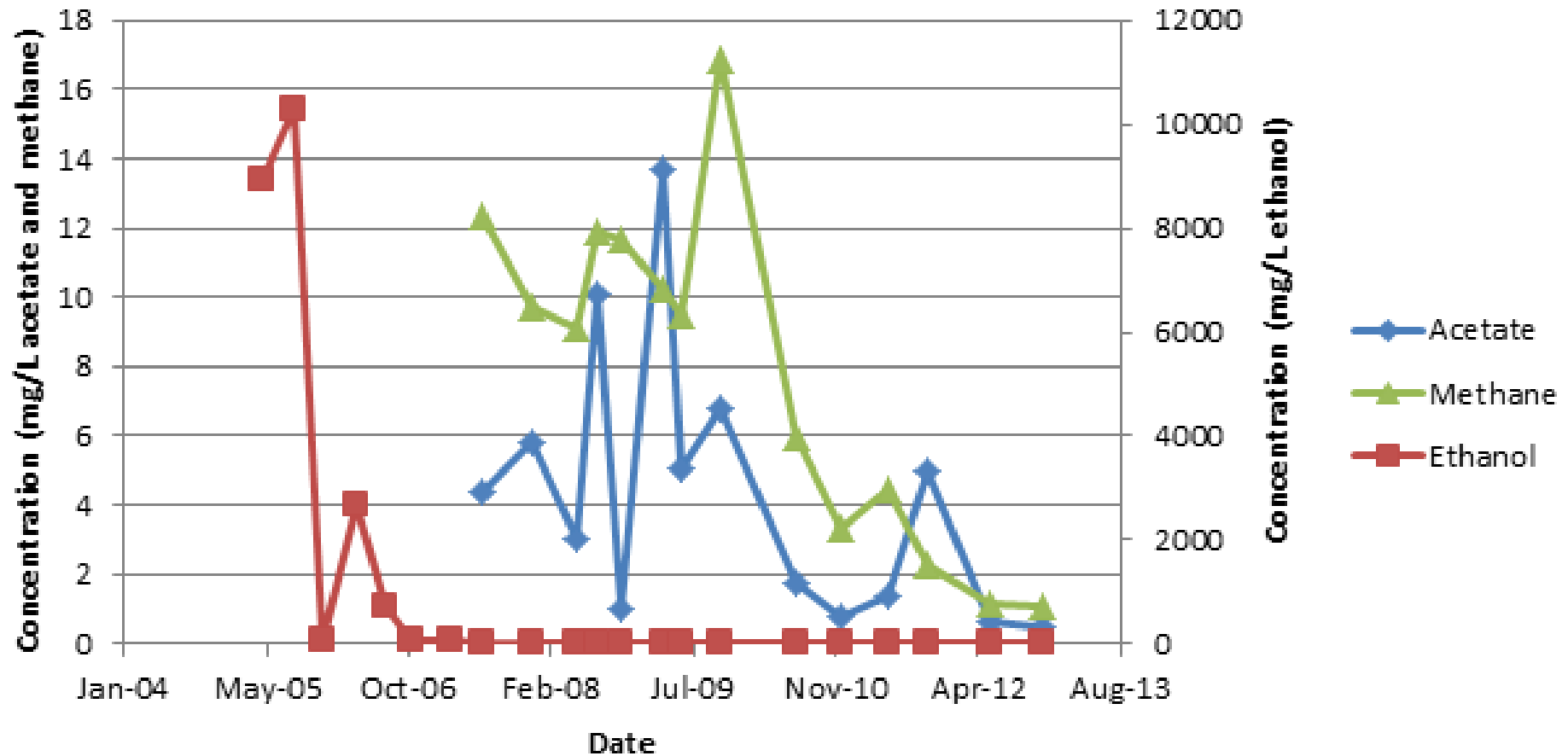


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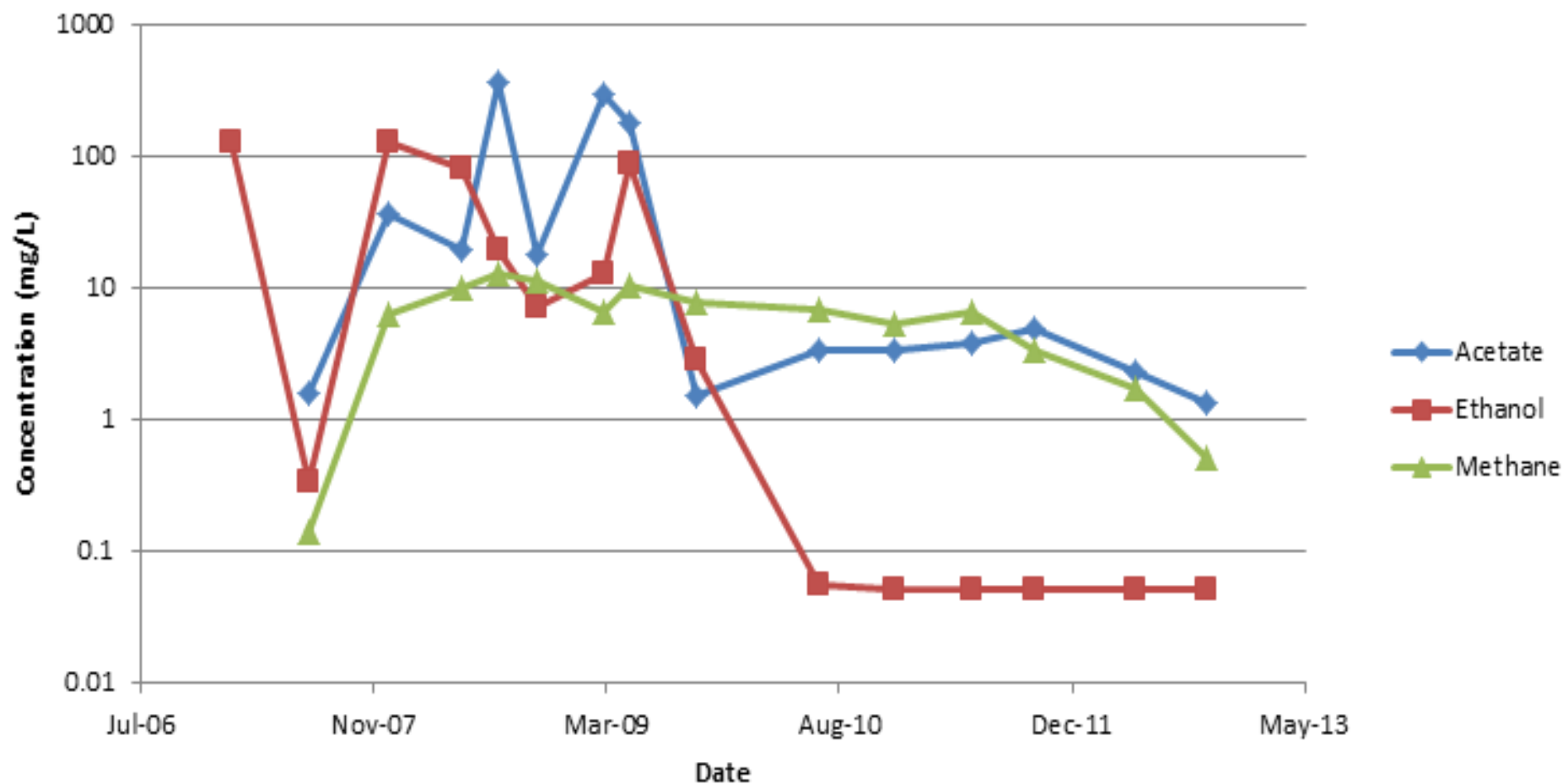
METHANE IN GROUNDWATER - OCTOBER 06, 2009
E-95 SPILL
CAMBRIA, MINNESOTA
FOR: MINNESOTA POLLUTION CONTROL AGENCY

Drawn:	TJC	4/23/10
Checked:	TJC	4/23/10
Approved:	RLD	4/23/10
PROJECT NUMBER	60154223	
FIGURE NUMBER	50	

Balaton Parameter Concentrations Over Time

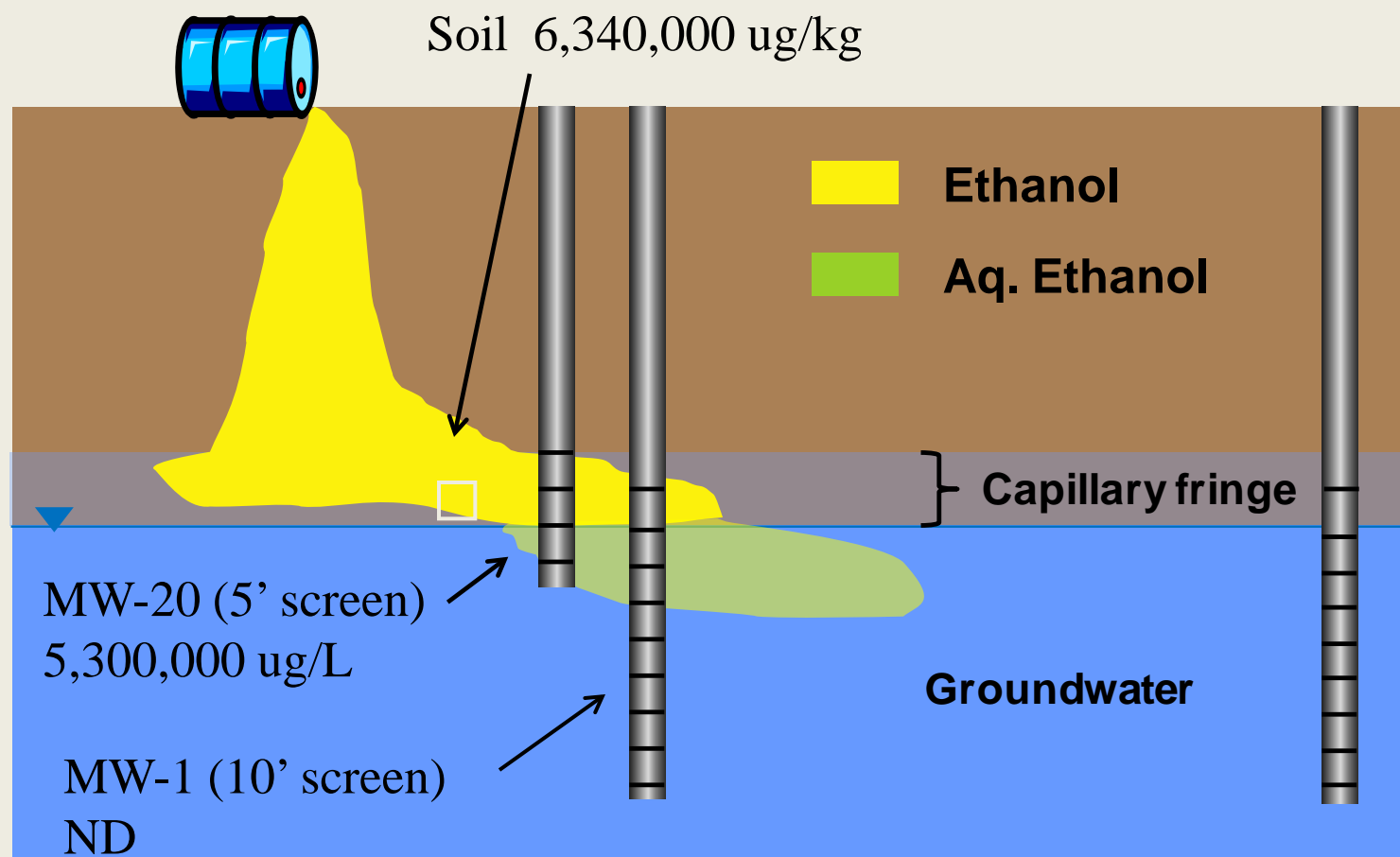


Cambria Parameter Concentrations Over Time



Cambria, MN

Oct 2011 Results

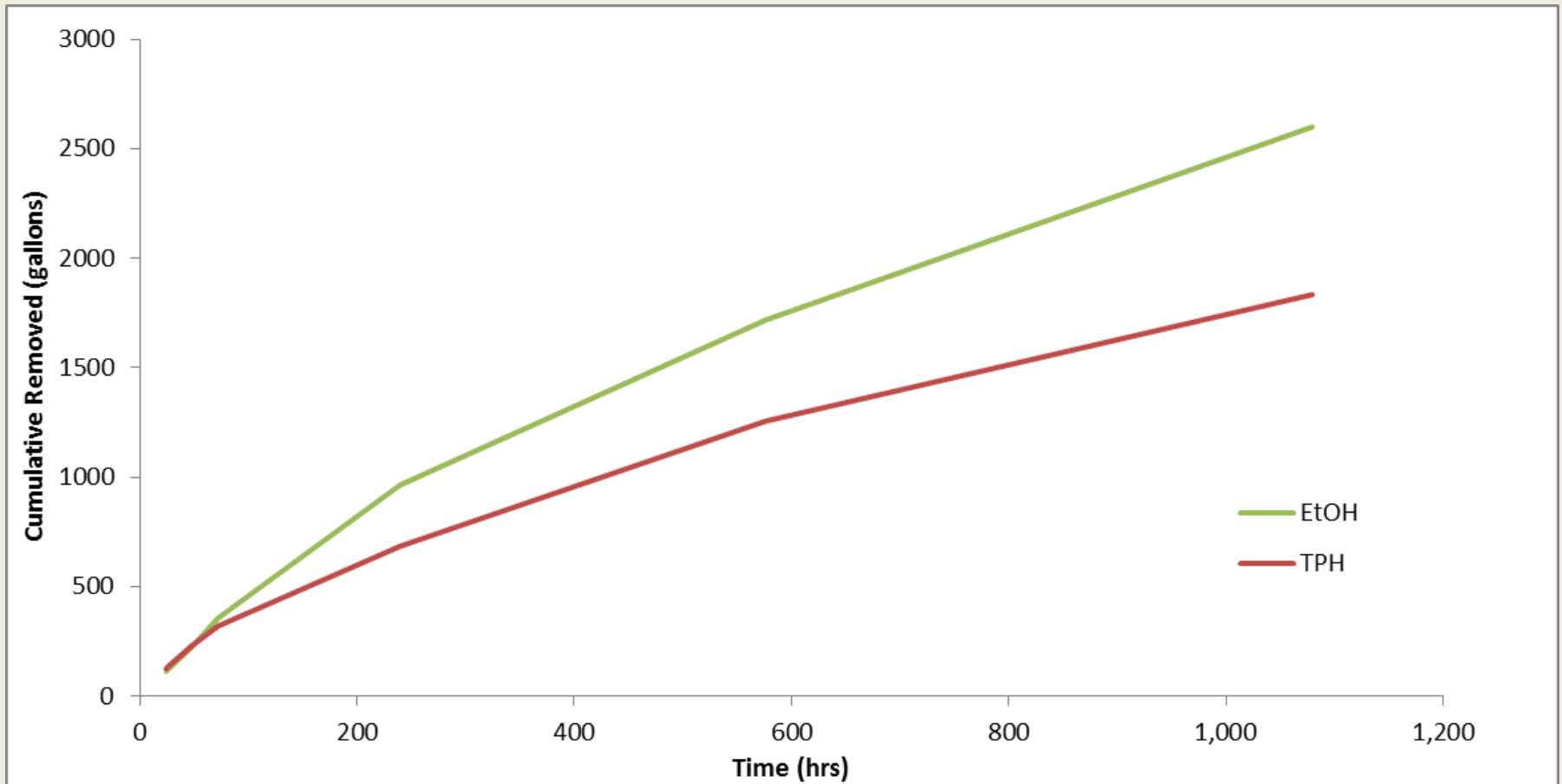


Remediation

- ❑ Simple - Just add (a lot) of oxygen!
 - Very few real world case studies
 - SVE, AS, Bioventing should work
- ❑ Methane risks can be remedied just like any other PVI issue
 - Sub-slab depressurization (radon) systems
- ❑ Groundwater recovery (PnT) can remove significant mass of ethanol if started immediately
 - Remove source, no ethanol, no methane



SVE Results, E85 Release



July 2008 – Lanesboro, MN

3,000 gals E95



References

Sihota, N.J., Mayer, K.U., Toso, M., and Atwater, J.A., 2013. “Methane emissions and contaminant degradation rates at sites affected by accidental releases of denatured fuel-grade ethanol.” J. Contam. Hydrol., 151 (1-15)

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Spalding, R.F., Toso, M.A., Exner, M.E., Hattan, G., Higgins, T.M., Sekely, A.C., Jensen, S.D., 2011. “Long-term groundwater monitoring results at large, sudden denatured ethanol releases”. Ground Water Monit. Rem. 31, 69–81.

Nelson, D., LaPara, T., Novak, P. (2010). “Effects of Ethanol-Based Fuel Contamination: Microbial Community Changes, Production of Regulated Compounds, and Methane Generation” Environ. Sci. Technol. 22(12), 4525-4530



Questions?

August 2014

430,000 gal DFE (E95) AST

~ 7000 gals released

Cause: Thin film epoxy coating failure, followed by stress corrosion cracking of steel floor plate

